

ORIGINAL ARTICLE

Design, synthesis and SAR study of hydroxychalcone inhibitors of human β -secretase (BACE1)

Lei Ma^{1,2}, Zhengyi Yang¹, Chenjing Li¹, Zhiyuan Zhu¹, Xu Shen¹, and Lihong Hu^{1,2}

¹Shanghai Research Center for Modernization of Traditional Chinese Medicine, Shanghai Institute of Materia Medica, Chinese Academy of Sciences, Shanghai, P. R. China, and ²School of Pharmacy, East China University of Science and Technology, Shanghai, P. R. China

Abstract

According to the structural characteristics of isoliquiritigenin from *Glycyrrhiza uralensis*, a series of hydroxychalcones has been designed, synthesized and evaluated for their *in vitro* inhibitory activities of β -secretase (BACE1). Structure-activity relationship study suggested that inhibitory activity against BACE1 was governed to a greater extent by the hydroxyl substituent on A- and B-ring of the chalcone, and the most active compound was substituted with four hydroxyl group (**17**, $IC_{50} = 0.27 \mu\text{M}$).

Keywords: Alzheimer's disease, BACE1, chalcone, SAR, *Glycyrrhiza uralensis*

Introduction

Alzheimer's disease (AD) is a devastating neurodegenerative disorder characterized by loss of memory and cognition. The accumulation of the β -amyloid peptide ($A\beta$) in the brain has been thought to be a key factor in the pathogenesis of the disease¹. $A\beta$ peptides are derived from a sequential proteolytic cleavage of amyloid-precursor protein (APP) by β -secretase (BACE1) and γ -secretase². The limiting step in this process is cleavage of membrane-bound APP by BACE1 to form soluble APP_{3–6}. So BACE1 is becoming a molecular target for therapeutic intervention in AD^{4,7–10}. And then numerous BACE1 inhibitors have been synthesized and tested, but most of reported inhibitors have focused on peptide-derived structures, which act as transition state analogs based on the amino acid sequences at the cleavage site of APP by BACE1¹¹. These species showed good activity *in vitro*, but their viability as drug candidates in this case is minimal because the enzyme inhibitors with therapeutic potential are preferably smaller than 700 Da, so large peptide-based inhibitors are not viable drug candidates. Thus, we tried to find the plant-derived non-peptidic

inhibitors against BACE1 which would be more likely to be able to reach the target.

With the goal of obtaining new natural inhibitors, we decided to investigate BACE1 inhibitors from the roots of *Glycyrrhiza uralensis* F., a traditional medicinal herb and an important perennial legume. This herb has long been valued as a demulcent, to relieve respiratory ailments, stomach burn including heart burn, gastritis, inflammatory disorders, skin diseases, and liver problems. Recently, it was reported that *G. uralensis* water extract markedly improved the cognitive deficits induced in mice by $A\beta_{25–35}$ administration¹². To look for the biologically active compounds conferring this activity, extract compounds were purified by conventional chromatography and the individual components were assessed for BACE1 activity using a BACE1 enzyme-based bioassay. Fortunately, we found isoliquiritigenin (2', 4', 4'-trihydroxychalcone, **1**) showed moderate BACE1 inhibitory activity ($IC_{50} = 33.0 \mu\text{M}$), and we also found **1** was structurally similar to the natural products catechins¹³ and hispidin¹⁴ which have also been reported to have BACE1 activity. In this paper, we describe the development of low molecular weight

Address for Correspondence: Lihong Hu, Shanghai Research Center for Modernization of Traditional Chinese Medicine, Shanghai Institute of Materia Medica, Chinese Academy of Sciences, Shanghai 201203, P. R. China. Tel: 86 21 50272221. Fax: 86 21 50272221. E-mail: simmhulh@mail.shnc.ac.cn.

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BACE1 inhibitors containing a chalcone skeleton through enzymatic assays.

Chemistry

The synthesis of targeted hydroxychalcone analogues (**2–22**) were achieved using commercially available hydroxyacetophenone and hydroxybenzaldehyde as starting materials, following route as shown in Scheme 1. Hydroxyacetophenone and hydroxybenzaldehyde was benzyl protected under basic conditions, and subsequently reacted in 40% aqueous solution of KOH to form the benzyl protected hydroxychalcone. Then deprotection of the benzyl ether was achieved by treatment with boron trichloride to produce the targeted hydroxychalcone analogues.

Results and discussion

Isoliquiritigenin was prepared to confirm the scaffold compound, and a series of derivatives were synthesized to evaluate their potential in terms of modification of the initial compound, to clarify their structure-activity relationships (SARs).

To determine the best substitution position of a hydroxyl group at B-ring, three compounds with the general structure shown in Table 1 were designed. The activity results for these compounds are listed in Table 1. It seems that BACE1 is sensitive to the position of the substituted group on the B-ring. The result for a hydroxyl substituted at position 2 was superior to those for the hydroxyl substituted at position 3 or position 4. These results indicate the B-ring of isoliquiritigenin (**1**) for modification to improve the activity of the inhibitor.

To further improve the inhibitory potency and determine the effect of the substituted group at position 2, four compounds (**3–6**) with different substituents at position 2 of B-ring were synthesized. The inhibitory activities of these compounds toward BACE1 are summarized in Table 2. To the position 2, the introduction of electron-donating substituents showed better activity than of electron-withdrawing substituents, and the compound **3** with the hydroxyl substituent had the best activity.

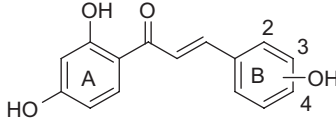
We next turned our attention to modify the A-ring of compound **3**. To determine the best substitution position

and number of hydroxyl groups at A-ring, nine compounds with the general structure shown in Table 3 were designed. The activity results for these compounds are listed in Table 3. It seems that BACE1 is also sensitive to the substituted positions and numbers of the hydroxyls on the A-ring. The tendency is obvious: the compounds with two hydroxyls (**3, 12–16**) showed higher inhibitory activity, and the compounds without or with only one hydroxyl (**8–11**) in ring A show unsatisfactory reduction of the activity of BACE1, so we choose **3**, the best activity compound as the beginning compound of the next structure modification.

To investigate the SAR of B-ring, six 2-hydroxyl-substituted compounds (Table 4) were designed. As can be seen from Table 1 and 4, the compounds with two hydroxyls (**17–22**) at B-ring showed better inhibitory activity than those with only one hydroxyl (**1–3**). Compound **17** with two hydroxyls substituted at the position 2 and 3 showed the best inhibitory activity ($IC_{50} = 0.27 \mu M$).

In summary, according to the SARs of all the 22 hydroxychalcones, we found that inhibitory activity against BACE1 was governed to a greater extent by the hydroxyl substituent on A- and B-ring of the chalcone, and the hydroxyl at the position 2, 2' and 4' of the chalcone maybe play a critical role in the activity.

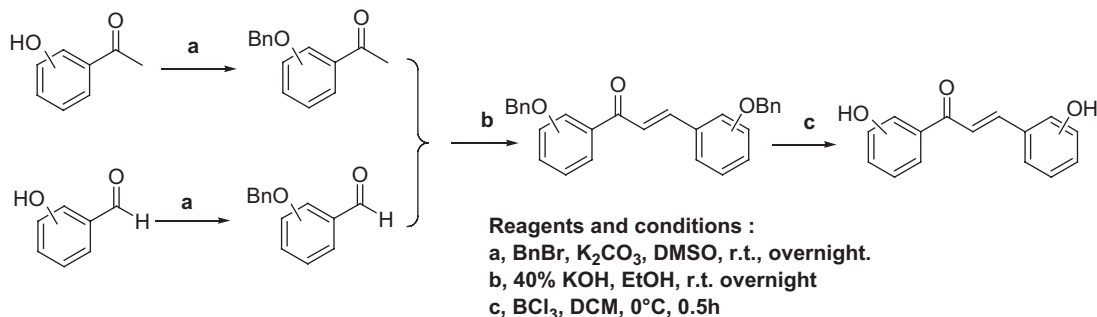
Table 1. Importance of the position of a hydroxyl substituent at B-ring.



Compound	Position of OH	IC_{50}^a (μM) to BACE1
1	Position 4	33.00 ± 3.75
2	Position 3	8.03 ± 0.92
3	Position 2	2.45 ± 0.39
HEA ^b		0.049 ± 0.011

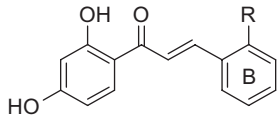
^aIn all tables, IC_{50} s reported are means of the values of three different experiments. Each IC_{50} is within threefold of the mean value.

^bHydroxyethylamine, the positive control in the assay demonstrated an IC_{50} value of $0.049 \pm 0.011 \mu M$, see ref. 16 for details.



Scheme 1. Preparation of hydroxychalcones.

Table 2. Diverse compounds at the position 2 of B-ring.



Compound	R	IC ₅₀ (μM) to BACE1
3	OH	2.45 ± 0.39
4	OMe	2.63 ± 0.28
5	OCH ₂ CH=CH ₂	>60
6	Cl	11.40 ± 1.36
7	NO ₂	>60

Experimental section

General experimental procedures

Unless otherwise mentioned, all chemicals and materials were used as received from commercial suppliers without further purification. Dichloromethane was distilled from calcium hydride. Thin-layer chromatography (TLC) plates (silica gel 60 GF, with glass support) from Yantai jiangyou company were used for monitoring progress of a reaction and visualized with 254 nm UV light. Nuclear magnetic resonance (NMR) spectra were recorded on a Bruker AM-400 spectrometer. EI-MS was obtained on a SHIMADZU GCMS-QP5050A spectrometer.

Extraction and isolation

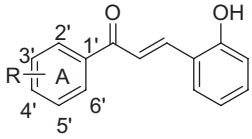
The air-dried and powdered roots of *G. uralensis* (2 kg) were percolated with 70% EtOH to afford 130 g of brown extract. Then the extract was suspended in water (500 mL) and further extracted with EtOAc to afford a brown semisolid (33 g after evaporation). The organic layer was subjected to CC (RP-18; MeOH/H₂O 60:40) to afford isoliquiritigenin (**1**) (320 mg).

General procedure for the preparation of compounds (2–22)

To a stirred solution of hydroxyacetophenone (1 mmol) or hydroxybenzaldehyde (0.1 mL, 1 mmol) in dimethyl sulfoxide (DMSO) (10 mL), was added K₂CO₃ 552 mg or 276 mg, respectively. Then, BnBr 0.2 mL (2 mmol) was added to the mixture, and the mixture was stirred at room temperature overnight. After completion of the reaction as indicated by TLC, the mixture was diluted with water and extracted with ethyl acetate (3 × 10 mL). The combined organic layer was washed with brine and dried over Na₂SO₄. Then, the organic layer was concentrated in a vacuum to afford the corresponding compound of benzyloxyacetophenone or benzyloxybenzaldehyde, which could be used directly for the next step without further purification.

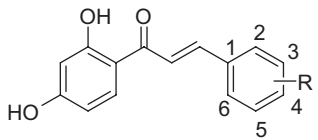
A 40% aqueous solution of potassium hydroxide (5 mL) was added drop-wise to the stirred solution of benzyloxyacetophenone (prepared by the first step) in ethanol (15 mL) under ice-bath. Then, a solution of benzyloxybenzaldehyde in ethanol (10 mL) was added drop-wise to the above mixture, and continued to stir

Table 3. Importance of the positions of hydroxyl substituents at A-ring.



Compound	Position of OH	IC ₅₀ (μM) to BACE1
3	Positions 2' and 4'	2.45 ± 0.39
8		>60
9	Position 2'	>60
10	Position 3'	30.4 ± 4.25
11	Position 4'	58.1 ± 4.92
12	Positions 2' and 3'	21.8 ± 2.31
13	Positions 2' and 5'	41.8 ± 5.52
14	Positions 2' and 6'	15.6 ± 1.83
15	Positions 3' and 4'	11.8 ± 1.35
16	Positions 3' and 5'	19.4 ± 2.33

Table 4. Importance of the positions of hydroxyl substituents at B-ring.



Compound	Position of OH	IC ₅₀ (μM) to BACE1
3	Position 2	2.45 ± 0.39
17	Positions 2 and 3	0.27 ± 0.05
18	Positions 2 and 4	0.62 ± 0.08
19	Positions 2 and 5	1.94 ± 0.20
20	Positions 2 and 6	1.71 ± 0.21
21	Positions 3 and 4	2.37 ± 0.34
22	Positions 3 and 5	2.08 ± 0.38

for 30 min under ice-bath, then stirred at room temperature overnight. After completion of the reaction as indicated by TLC, the mixture was concentrated in a vacuum and the residue was diluted with water and extracted with ethyl acetate (3 × 15 mL). The combined organic layer was washed with brine, dried over Na₂SO₄, and concentrated in a vacuum, the residue was chromatographed on silica gel to afford benzyloxy-chalcone.

To a stirred solution of benzyloxy-chalcone in anhydrous dichloromethane (15 mL) was added drop-wise with 5 mL of BCl₃ (1 M in dichloromethane) under ice-bath and the mixture was stirred for 30 min under the same condition. Then ice water (15 mL) was added to the mixture and stirred at room temperature for another 2.5 h. After completion of the reaction as indicated by TLC, the mixture was concentrated in a vacuum and the residue was extracted with ethyl acetate (3 × 15 mL). The combined organic layer was washed with brine, dried over Na₂SO₄, and concentrated in a vacuum, and the residue was chromatographed on silica gel to afford hydroxylchalcone (**2–22**).

(E)-1-(2,4-dihydroxyphenyl)-3-(3-hydroxyphenyl)-2-propen-1-one (2)

Yield 82%, yellow solid. Proton-NMR (^1H NMR) ($\text{DMSO-}d_6$, 400 MHz): δ : 13.4 (1 H, br s), 10.8 (1 H, br s), 9.64 (1 H, br s), 8.19 (1 H, d, $J=8.7$), 7.89 (1 H, d, $J=15.6$), 7.70 (1 H, d, $J=15.6$), 7.23–7.34 (3 H, m), 6.87 (1 H, d, $J=7.5$), 6.42 (1 H, d, $J=8.7$), 6.30 (1 H, d, $J=2.1$); HR-EI-MS: 256.0730 (M^+ , $\text{C}_{15}\text{H}_{12}\text{O}_4^+$; calc. 256.0736).

(E)-1-(2,4-dihydroxyphenyl)-3-(2-hydroxyphenyl)-2-propen-1-one (3)

^1H NMR ($\text{DMSO-}d_6$, 400 MHz): δ : 13.52 (1 H, s, OH-2), 10.72 (1 H, br s, OH-4), 10.31 (1 H, br s, OH-2'), 8.07 (1 H, d, $J=15.6$, H- α), 7.74 (1 H, d, $J=15.6$, H- β), 7.78 (1 H, d, $J=8.7$, H-6'), 7.51 (1 H, dd, $J=8.7$, 2.1, H-6), 7.19 (1 H, td, $J=8.7$, 2.7, H-5), 6.85 (1 H, td, $J=8.7$, 2.4, H-4), 6.83 (1 H, td, $J=8.7$, 2.7, H-3), 6.39 (1 H, dd, $J=8.7$, 2.4, H-5'), 6.32 (1 H, d, $J=2.4$, H-3'); HR-EI-MS: 256.0731 (M^+ , $\text{C}_{15}\text{H}_{12}\text{O}_4^+$; calc. 256.0736).

(E)-1-(2,4-dihydroxyphenyl)-3-(2-methoxyphenyl)-2-propen-1-one (4)

Yield 83%, yellow solid. ^1H NMR ($\text{DMSO-}d_6$, 400 MHz): δ : 13.5 (1 H, br s), 8.19 (1 H, d, $J=15.6$, H- α), 7.84 (1 H, d, $J=9.6$), 7.69 (1 H, d, $J=15.6$, H- β), 7.63 (1 H, d, $J=7.5$), 6.40 (1 H, t, $J=7.5$), 7.00 (1 H, d, $J=7.8$), 6.96 (1 H, d, $J=8.7$), 6.41–6.45 (2 H, m), 3.94 (3 H, s); HR-EI-MS: 270.0899 (M^+ , $\text{C}_{16}\text{H}_{14}\text{O}_4^+$; calc. 270.0892).

(E)-1-(2,4-dihydroxyphenyl)-3-(2-allyloxyphenyl)-2-propen-1-one (5)

Yield 23%, yellow solid. ^1H NMR ($\text{DMSO-}d_6$, 400 MHz): δ : 13.5 (1 H, br s), 10.7 (1 H, br s), 9.28 (1 H, br s), 8.24 (1 H, d, $J=15.0$, H- α), 8.16 (1 H, d, $J=8.4$), 7.87 (1 H, d, $J=15.6$, H- β), 7.83 (1 H, d, $J=7.2$), 7.17 (1 H, d, $J=7.2$), 6.90 (1 H, t, $J=7.8$), 6.41 (1 H, d, $J=9.3$), 6.29 (1 H, d, $J=2.4$), 5.89–6.00 (1 H, m), 5.06 (1 H, dd, $J=3.3$, 1.5), 5.02 (1 H, d, $J=1.2$), 3.39 (2 H, d, $J=5.7$); HR-EI-MS: 296.1046 (M^+ , $\text{C}_{18}\text{H}_{16}\text{O}_4^+$; calc. 296.1049).

(E)-1-(2,4-dihydroxyphenyl)-3-(2-chlorophenyl)-2-propen-1-one (6)

Yield 85%, yellow solid. ^1H NMR ($\text{DMSO-}d_6$, 400 MHz): δ : 13.2 (1 H, br s), 10.8 (1 H, br s), 8.19–8.25 (2 H, m), 8.10 (1 H, d, $J=15.6$, H- α), 8.03 (1 H, d, $J=15.6$, H- β), 7.56–7.60 (1 H, m), 7.45–7.51 (2 H, m), 6.43 (1 H, dd, $J=8.7$, 2.1), 6.31 (1 H, d, $J=2.7$); HR-EI-MS: 274.0391 (M^+ , $\text{C}_{15}\text{H}_{11}\text{ClO}_3^+$; calc. 274.0397).

(E)-1-(2,4-dihydroxyphenyl)-3-(2-nitrophenyl)-2-propen-1-one (7)

Yield 77%, yellow solid. ^1H NMR ($\text{DMSO-}d_6$, 400 MHz): δ : 13.1 (1 H, br s), 10.9 (1 H, br s), 8.20 (2 H, t, $J=6.6$), 8.10 (1 H, dd, $J=4.8$, 1.5), 8.00 (1 H, d, $J=4.2$), 7.83 (1 H, td, $J=5.1$, 1.2), 7.70 (1 H, td, $J=5.7$, 1.5), 6.45 (1 H, d, $J=5.7$), 6.34 (1 H, d, $J=1.2$); HR-EI-MS: 285.0642 (M^+ , $\text{C}_{15}\text{H}_{11}\text{NO}_5^+$; calc. 285.0637).

(E)-1-phenyl-3-(2-hydroxyphenyl)-2-propen-1-one (8)

Yield 81%, yellow solid. ^1H NMR (CDCl_3 , 400 MHz): δ : 8.16 (1 H, d, $J=15.6$, H- α), 8.04 (2 H, d, $J=6.9$), 7.70 (1 H, d, $J=15.6$, H- β), 7.60 (1 H, d, $J=7.2$), 7.58 (1 H, d, $J=6.6$), 7.52 (1 H, d, $J=7.5$), 7.49 (1 H, d, $J=6.9$), 7.28 (1 H, t, $J=7.5$), 6.96 (1 H, t, $J=7.5$), 6.91 (1 H, d, $J=8.1$), 6.56 (1 H, br s, OH-2'); HR-EI-MS: 224.0833 (M^+ , $\text{C}_{15}\text{H}_{12}\text{O}_2^+$; calc. 224.0837).

(E)-1-(2-hydroxyphenyl)-3-(2-hydroxyphenyl)-2-propen-1-one (9)

Yield 73%, yellow solid. ^1H NMR (CDCl_3 , 400 MHz): δ : 8.20 (1 H, d, $J=15.6$, H- α), 7.94 (1 H, d, $J=7.8$), 7.85 (1 H, d, $J=15.6$, H- β), 7.61 (1 H, d, $J=7.5$), 7.50 (1 H, t, $J=7.2$), 7.30 (1 H, d, $J=7.2$), 7.03 (1 H, d, $J=7.2$), 6.98 (1 H, d, $J=8.1$), 6.93 (1 H, d, $J=8.1$), 6.85 (1 H, d, $J=8.4$), 5.73 (1 H, br s); HR-EI-MS: 240.0791 (M^+ , $\text{C}_{15}\text{H}_{12}\text{O}_3^+$; calc. 240.0786).

(E)-1-(3-hydroxyphenyl)-3-(2-hydroxyphenyl)-2-propen-1-one (10)

Yield 75%, yellow solid. ^1H NMR ($\text{DMSO-}d_6$, 400 MHz): δ : 10.3 (1 H, br s), 9.79 (1 H, br s), 8.00 (1 H, d, $J=15.6$, H- α), 7.83 (1 H, dd, $J=5.2$, 1.2), 7.77 (1 H, d, $J=15.6$, H- β), 7.25 (1 H, dt, $J=5.2$, 0.9), 7.40 (1 H, m), 7.36 (1 H, t, $J=7.8$), 7.27 (1 H, td, $J=7.5$, 1.5), 7.04 (1 H, ddd, $J=7.9$, 2.4, 0.9), 6.93 (1 H, dd, $J=7.9$, 0.9), 6.87 (1 H, t, $J=7.2$); HR-EI-MS: 240.0790 (M^+ , $\text{C}_{15}\text{H}_{12}\text{O}_3^+$; calc. 240.0786).

(E)-1-(4-hydroxyphenyl)-3-(2-hydroxyphenyl)-2-propen-1-one (11)

Yield 71%, yellow solid. ^1H NMR ($\text{DMSO-}d_6$, 400 MHz): δ : 10.4 (1 H, br s), 10.2 (1 H, br s), 8.01 (2 H, d, $J=8.4$), 7.98 (1 H, d, $J=15.6$, H- α), 7.84 (1 H, d, $J=5.4$), 7.82 (1 H, d, $J=15.6$, H- β), 7.25 (1 H, td, $J=7.8$, 1.2), 6.84–6.94 (4 H, m); HR-EI-MS: 240.0793 (M^+ , $\text{C}_{15}\text{H}_{12}\text{O}_3^+$; calc. 240.0786).

(E)-1-(2,3-dihydroxyphenyl)-3-(2-hydroxyphenyl)-2-propen-1-one (12)

Yield 61%, yellow solid. ^1H NMR ($\text{DMSO-}d_6$, 400 MHz): δ : 12.7 (1 H, br s), 10.3 (1 H, br s), 9.37 (1 H, br s), 8.16 (1 H, d, $J=15.6$, H- α), 7.89 (1 H, d, $J=9.9$), 7.93 (1 H, d, $J=15.6$, H- β), 7.63 (1 H, d, $J=8.1$), 7.30 (1 H, t, $J=6.9$), 7.07 (1 H, d, $J=7.8$), 6.95 (1 H, d, $J=8.1$), 6.89 (1 H, t, $J=7.8$), 6.82 (1 H, t, $J=7.8$); HR-EI-MS: 256.0733 (M^+ , $\text{C}_{15}\text{H}_{12}\text{O}_4^+$; calc. 256.0736).

(E)-1-(2,5-dihydroxyphenyl)-3-(2-hydroxyphenyl)-2-propen-1-one (13)

Yield 68%, yellow solid. ^1H NMR ($\text{DMSO-}d_6$, 400 MHz): δ : 11.9 (1 H, br s), 10.4 (1 H, br s), 9.20 (1 H, br s), 8.08 (1 H, d, $J=15.6$, H- α), 7.82 (1 H, d, $J=8.7$), 7.86 (1 H, d, $J=15.6$, H- β), 7.42 (1 H, d, $J=2.7$), 7.29 (1 H, t, $J=7.8$), 7.02 (1 H, dd, $J=9.0$, 2.7), 6.82–6.96 (3 H, m); HR-EI-MS: 256.0732 (M^+ , $\text{C}_{15}\text{H}_{12}\text{O}_4^+$; calc. 256.0736).

(E)-1-(2,6-dihydroxyphenyl)-3-(2-hydroxyphenyl)-2-propen-1-one (14)

Yield 63%, yellow solid. ^1H NMR ($\text{DMSO-}d_6$, 400 MHz): δ : 11.2 (2 H, br s), 10.2 (1 H, br s), 7.89 (1 H, d, $J=15.6$, H- α),

7.58 (1 H, d, $J=7.8$), 7.79 (1 H, d, $J=15.6$, H- β), 7.26 (1 H, t, $J=7.2$), 7.20 (1 H, t, $J=8.1$), 6.92 (1 H, d, $J=8.1$), 6.86 (1 H, t, $J=7.8$), 6.38 (2 H, d, $J=8.1$); HR-EI-MS: 256.0729 (M^+ , $C_{15}H_{12}O_4^+$; calc. 256.0736).

(E)-1-(3,4-dihydroxyphenyl)-3-(2-hydroxyphenyl)-2-propen-1-one (15)

Yield 59%, yellow solid. 1H NMR (DMSO- d_6 , 400 MHz): δ : 10.2 (2 H, br s), 9.91 (1 H, br s), 9.39 (1 H, br s), 7.94 (1 H, d, $J=15.6$, H- α), 7.81 (1 H, d, $J=6.3$), 7.78 (1 H, d, $J=15.6$, H- β), 7.56 (1 H, d, $J=8.4$), 7.48 (1 H, br s), 7.25 (1 H, d, $J=7.2$), 6.92 (1 H, t, $J=8.1$), 6.87 (1 H, d, $J=5.7$), 6.85 (1 H, d, $J=8.1$); HR-EI-MS: 256.0733 (M^+ , $C_{15}H_{12}O_4^+$; calc. 256.0736).

(E)-1-(3,5-dihydroxyphenyl)-3-(2-hydroxyphenyl)-2-propen-1-one (16)

Yield 65%, yellow solid. 1H NMR (DMSO- d_6 , 400 MHz): δ : 10.2 (1 H, br s), 9.60 (2 H, br s), 7.93 (1 H, d, $J=15.6$, H- α), 7.77 (1 H, d, $J=7.8$), 7.66 (1 H, d, $J=15.6$, H- β), 7.26 (1 H, t, $J=6.9$), 6.84–6.98 (4 H, m), 6.47 (1 H, br s); HR-EI-MS: 256.0738 (M^+ , $C_{15}H_{12}O_4^+$; calc. 256.0736).

(E)-1-(2,4-dihydroxyphenyl)-3-(2,3-dihydroxyphenyl)-2-propen-1-one (17)

Yield 54%, yellow solid. 1H NMR (DMSO- d_6 , 400 MHz): δ : 13.5 (1 H, br s), 8.15 (1 H, d, $J=15.6$, H- α), 8.09 (1 H, d, $J=9.0$), 7.83 (1 H, d, $J=15.6$, H- β), 7.35 (1 H, d, $J=7.8$), 6.87 (1 H, d, $J=7.8$), 6.71 (1 H, t, $J=7.8$), 6.41 (1 H, dd, $J=9.0$, 1.5), 6.29 (1 H, d, $J=1.2$); EI-MS: 272 [M] $^+$.

(E)-1-(2,4-dihydroxyphenyl)-3-(2,4-dihydroxyphenyl)-2-propen-1-one (18)

Yield 51%, yellow solid. 1H NMR (DMSO- d_6 , 400 MHz): δ : 13.8 (1 H, br s), 8.07 (1 H, d, $J=15.6$, H- α), 8.05 (1 H, d, $J=9.0$), 7.68 (1 H, d, $J=15.6$, H- β), 7.73 (1 H, d, $J=8.4$), 6.37–6.41 (2 H, m), 6.32 (1 H, dd, $J=7.2$, 2.4), 6.26 (1 H, d, $J=2.1$); HR-EI-MS: 272.0681 (M^+ , $C_{15}H_{12}O_5^+$; calc. 272.0685).

(E)-1-(2,4-dihydroxyphenyl)-3-(2,5-dihydroxyphenyl)-2-propen-1-one (19)

Yield 47%, yellow solid. 1H NMR (DMSO- d_6 , 400 MHz): δ : 13.5 (1 H, br s), 10.7 (1 H, br s), 9.61 (1 H, br s), 8.91 (1 H, br s), 8.05 (1 H, d, $J=15.6$, H- α), 8.09 (1 H, d, $J=8.4$), 7.77 (1 H, d, $J=15.6$, H- β), 7.22 (1 H, br s), 6.75 (1 H, d, $J=0.9$), 6.41 (1 H, dd, $J=9.0$, 2.4), 6.28 (1 H, d, $J=2.7$); HR-EI-MS: 272.0682 (M^+ , $C_{15}H_{12}O_5^+$; calc. 272.0685).

(E)-1-(2,4-dihydroxyphenyl)-3-(2,6-dihydroxyphenyl)-2-propen-1-one (20)

Yield 49%, yellow solid. 1H NMR (DMSO- d_6 , 400 MHz): δ : 8.45 (1 H, d, $J=15.6$, H- α), 7.83 (1 H, d, $J=9.0$), 8.25 (1 H, d, $J=15.6$, H- β), 7.03 (1 H, t, $J=8.1$), 6.50 (1 H, d, $J=8.1$), 6.46 (1 H, dd, $J=9.0$, 2.1), 6.36 (1 H, d, $J=2.7$); HR-EI-MS: 272.0678 (M^+ , $C_{15}H_{12}O_5^+$; calc. 272.0685).

(E)-1-(2,4-dihydroxyphenyl)-3-(3,4-dihydroxyphenyl)-2-propen-1-one (21)

Yield 49%, yellow solid. 1H NMR (DMSO- d_6 , 400 MHz): δ : 7.73 (1 H, d, $J=8.4$), 7.65 (1 H, d, $J=15.3$), 7.32 (1 H, d, $J=15.3$), 7.08 (1 H, s), 7.00 (1 H, d, $J=8.4$), 6.76 (1 H, d, $J=8.4$), 6.33 (1 H, dd, $J=8.4$, 1.8), 6.28 (1 H, d, $J=1.8$); HR-EI-MS: 272.0688 (M^+ , $C_{15}H_{12}O_5^+$; calc. 272.0685).

(E)-1-(2,4-dihydroxyphenyl)-3-(3,5-dihydroxyphenyl)-2-propen-1-one (22)

Yield 55%, yellow solid. 1H NMR (DMSO- d_6 , 400 MHz): δ : 13.4 (1 H, br s), 10.7 (1 H, br s), 9.49 (2 H, br s), 7.78 (1 H, d, $J=15.6$, H- α), 8.17 (1 H, d, $J=9.0$), 7.58 (1 H, d, $J=15.6$, H- β), 6.70 (2 H, s), 6.28–6.42 (3 H, m); HR-EI-MS: 272.0687 (M^+ , $C_{15}H_{12}O_5^+$; calc. 272.0685).

Biochemical assay

The biochemical assay was employed to determine the IC_{50} of compounds. The assay based on fluorescence-resonance energy transfer was carried out with BACE1 enzyme at pH 4.5 with a substrate, H-Lys(DABSYL)-SEVNLDAEFR-Gin-(LY) (Merck) according to Ermolieff's approach¹⁵. Briefly, the assay was carried out by incubation of the substrate peptide (5 μ M), BACE1 enzyme (550 ng/mL) that was expressed and purified from insect expression system and varied concentrations of compounds at pH 4.5. After 1 h incubation at 37°C, stop buffer (3 M sodium acetate) was introduced into the reaction mixture to stop the reaction. Finally, the fluorescence intensity was measured on a TECAN GENios reader (Tecan) at room temperature with excitation at 420 nm and emission at 530 nm.

Declaration of interest

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